ANTIOXIDANT ACTIVITY OF *ABUTILON INDICUM* LEAVES
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ABSTRACT: Antioxidant activity of methanolic extract of *Abutilon indicum* leaves was investigated for its free radical scavenging activity by determining the nitric oxide and superoxide radical scavenging activity. Maximum scavenging of nitric oxide and superoxide radical found were 28.74 % and 49.62 % respectively at 250 µg/ml concentration.

Keywords: *Abutilon indicum*, antioxidant, free radical, nitric oxide

INTRODUCTION
The plant kingdom represents an enormous reservoir of pharmacological molecules to be discovered⁴. Compounds of natural origin play a major role as drugs and as lead structures for the development of synthetic molecules. Natural products are being widely used in the form of medicinal plants. This scenario will definitely demand a revised Research and Development agenda for herbal medicines to be in tune to penetrate an estimated 40% of the synthetic pharmaceutical market by 2010 ². Plants are the essential and integral part in Complementary and Alternative medicine and due to this they develop the ability for the formation of secondary metabolites like proteins, flavonoids, alkaloids, steroids and phenolic substances which are in turn used to restore health and heal many diseases⁵. Thus the present investigation was aimed at evaluating the antioxidant activity of *Abutilon indicum* (Linn.) leaves. *Abutilon indicum* (Linn) family Malvaceae, commonly known as Atibala is an important medicinal plant used in our Traditional System of Medicine to treat various health ailments. The plant is used as demulcent, aphrodisiac, laxative, diuretic, pulmonary and sedative. The leaves are used as astringent ⁴, bark is used as diuretic and seeds are used as laxative, expectorant and demulcent ⁵. The plant contains mucilage, tannins, asparagines, gallic acid and sesquiterpenes ⁶.

EXPERIMENTAL
MATERIALS AND METHODS
PLANT MATERIAL: Fresh leaves of the plant *Abutilon indicum* (Linn.) were obtained and identified from authentic sources. A voucher specimen has been identified and deposited at the Department of Pharmacognosy, School of Pharmacy and Technology Management, Shirpur. The collected leaves were dried in shade, crushed to coarse powder and used for further studies.

PREPARATION OF EXTRACT: The dried plant material leaves (500 gm) were subjected to continuous hot extraction with chloroform for 48 h in a soxhlet apparatus. The chloroform extract was filtered and partitioned by using petroleum ether to remove the fixed oils, fats and other non polar constituents present in it. The solvent was evaporated under reduced pressure and dried in a vacuum desiccator and was used when required. The dried extract (CEAI) thus obtained (280 gm) was used for the assessment of antioxidant activity. 36 hours ⁷.

The extracts were subjected to preliminary qualitative tests to identify the various phytoconstituents present in leaves ⁸. The qualitative chemical tests performed were Shinoda test, ammonia fuming test, lead acetate, boric acid for flavonoid containing compounds and ferric chloride test, nitric acid test, ammonia hydroxide – potassium ferricyanide test, lead acetate test for the presence of tannins. All these test gave positive results when they were compared with Rutin and Quercetin, standard drugs of the class.

NITRIC OXIDE RADICAL INHIBITION ASSAY
Nitric oxide radical inhibition was estimated by the use of Griess Illosvoy reaction ⁹. In this investigation, Griess Illosvoy reagent was generally modified by using naphthyl ethylene diamine dihydrochloride (0.1 % w/v) instead of the use of 1- naphthylamine (5 %). The
reaction mixture (3 ml) containing sodium nitroprusside (10 mM, 2 ml), phosphate buffer saline (0.5 ml) and the extract (20-250 µg/ml) standard solution (rutin, 0.5 ml) was incubated at 25°C for 150 minutes. A control test compound equivalent amount of methanol was taken. After incubation, 0.5 ml of the reaction mixture mixed with 1 ml sulfanilic acid reagent (0.33 % in 20 % glacial acetic acid) and allowed to stand for 5 min for completion of the reaction process of diazotization. Further, 1 ml of the naphthyl ethylene diamine dihydrochloride was added, mixed and was allowed to stand for 30 min at 25°C. The concentration of nitrite was assayed at 540 nm and was calculated with the reference to the absorbance of the standard nitrite solutions. Rutin was taken as a standard. The percent inhibition was calculated using the formula: % inhibition = [(A<sub>cont</sub> - A<sub>test</sub>) / A<sub>cont</sub>] X 100 ……(1). Where A<sub>cont</sub> is the absorbance of the control reaction and A<sub>test</sub> is the absorbance in the presence of samples with the extracts.

**SUPEROXIDE RADICAL SCAVENGING ACTIVITY**

Superoxide radical scavenging activity is generally based on the anion radical which is associated with PMS-NADH system. The measurement of superoxide scavenging activity of the is based on method as described by Liu<sup>10</sup>. They are generated within PMS-NADH systems by the oxidation of NADH and are assayed by the reduction of nitroblue tetrazolium (NBT). Tris HCl buffer (3 ml, 16 mM, pH 8.0) containing 1 ml NBT (50 µM) solution, 1 ml NADH (78 µM) solution and a sample solution of extract (20-250 µg/ml) in water were mixed. The reaction was started when 1 ml of phenazine methosulfate (PMS) solution (10 µM) was added to the mixture. The reaction mixture was incubated at 25°C for 5 min, and the absorbance was read at 560 nm against the corresponding blank samples. Quercetin was used as a reference drug. Decreased absorbance of the reaction mixture indicated increased superoxide anion scavenging activity. The percentage inhibition was calculated by using the same formula (1). Chloroform extract was dried under reduced vacuum pressure and finally it was made soluble in the corresponding solvents (phosphate-buffer). The tests were performed with the filtrate soluble portion.

**RESULTS AND DISCUSSION**

The preliminary qualitative tests indicated the presence of flavonoids, phenols and tannins. Table 1 and Table 2 shows that the percentage inhibition of nitric oxide and superoxide radical by AICE. The percentage inhibition for the superoxide radical was found to be moderate and nitric oxide radical is less significant when compared to the reference standard. Values are expressed as mean ± SEM of five measurements. Statistical analysis was performed by Dunnett’s Test by ANOVA. IC 50 values for all the above experiments were determined by linear regression analysis. The activity is increasing with the concentration and difference were statistically significant (p<0.01). After 250µg/ml there was decrease in the activity with a negative effect.

**DISCUSSION**

Oxidative stress in large quantities of reactive oxygen species (ROS) are generated is one of the earliest responses to stress. The antioxidant system protects the pathogens against the ROS-induced oxidative damage. Nitric oxide radical generated from the sodium nitropruside and measured by the Greiss reduction. Sodium nitropruside at physiological pH spontaneously generates nitric oxide, which thereby interacts with oxygen to produce nitrate ions that can be estimated by use of Greiss reagents. Thus the scavengers of nitric oxide compete with the oxygen, leading to reduced production of nitric oxide. In the PMS/NADH coupling reaction reduces NBT. The decrease of absorbance at 560 nm with antioxidants thus indicates the consumption of superoxide anion in the reaction mixture. Thus, AICE fraction, possessed good antioxidant property so, work should be carried out with more antioxidant models. The results were positive at the laboratory level and further work can be carried out to find out the exact constituent responsible for these activities by the process of modern analytical tools. Thus the plant can be a potential source for antioxidant property.

<table>
<thead>
<tr>
<th>Concentration (µg/ml)</th>
<th>Rutin (Std. %)</th>
<th>AICE (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>20</td>
<td>29.03 ± 0.43</td>
<td>7.75 ± 0.21</td>
</tr>
<tr>
<td>40</td>
<td>45.2± 1.4</td>
<td>12.02 ± 1.00</td>
</tr>
<tr>
<td>100</td>
<td>5424 ± 0.78</td>
<td>18.44 ± 1.13</td>
</tr>
<tr>
<td>125</td>
<td>68.2 ± 1.16</td>
<td>22.43 ±2.22</td>
</tr>
<tr>
<td>250</td>
<td>76.3 ± 0.42</td>
<td>28.74 ± 1.67</td>
</tr>
<tr>
<td>IC&lt;sub&gt;50&lt;/sub&gt; (µg/ml)</td>
<td>67.8</td>
<td>512.40</td>
</tr>
</tbody>
</table>

Values are presented as the mean±SEM (n=08)

AICE- Abutilon indicum chloroform extract; Std.- Standard.
Table 2. Antioxidant activity of chloroform extract of *Abutilon indicum* leaves

<table>
<thead>
<tr>
<th>Concentration (µg/ml)</th>
<th>Quercetin (Std. %)</th>
<th>AICE (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>20</td>
<td>32.0 ± 0.32</td>
<td>16.75 ± 0.44</td>
</tr>
<tr>
<td>40</td>
<td>51.2 ± 0.51</td>
<td>29.0 ± 1.02</td>
</tr>
<tr>
<td>100</td>
<td>57.3 ± 0.86</td>
<td>34.6 ± 1.12</td>
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<tr>
<td>125</td>
<td>58.5 ± 1.16</td>
<td>43.7 ± 2.64</td>
</tr>
<tr>
<td>250</td>
<td>63.3 ± 1.12</td>
<td>49.6 ± 1.71</td>
</tr>
<tr>
<td>IC&lt;sub&gt;50&lt;/sub&gt; (µg/ml)</td>
<td>62.5</td>
<td>235.4</td>
</tr>
</tbody>
</table>

* Values are presented as the mean±SEM (n=08)
AICE= *Abutilon indicum* chloroform extract; Std.- Standard.

REFERENCES