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Solvent extraction and spectrophotometric determination of Fe(III) by using5-bromo salicylidene-2-aminothiophenol (BSATP) as an analytical reagent

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Abstract:A spectrophotometric method has been developed for the determination of Fe (III) using5-bromo salicylidene-2-aminothiophenol (BSATP) as an extractive reagent.

The reagent forms a colored complex which has been quantitatively extracted into chloroform at pH 8.2. The method obeys Beer's law over a range of 1 to 10 ppm. The molar absorptivity is 16460 L mol⁻¹cm⁻¹ and Sandell'ssensitivity is $0.01639\mu g$ cm⁻² respectively. The proposed method is very sensitive and selective. This method has been successfully applied to synthetic and commercial samples.

Key words: Spectrophotometric determination, 5-bromo salicylidene-2-aminothiophenol (BSATP), chloroform, molar absorptivity.

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