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An easy synthesis of *progesterone-dihydropyrimidine derivative*

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Abstract: In this study, was synthesized a new *steroid-dihydropyrimidine derivative*; the route involved preparation of *progesterone-dihydropyrimidine derivative* (4) using the three components system (4-pregnene-3,20-dione (1), *benzaldheide* (2) and *thiourea* (3) in presence of *chloridric acid* such as catalyst. **Keywords.** *Progesterone, dihydropyrimidine, benzaldheide*, derivative.

Introduction

Combinatorial chemistry is a powerful tool for development of new drugs. In this context, over the past decade, several dihydropyrimidine-derivatives were synthesized with a wide spectrum of biological actions¹, as antibacterials^{2,3}, antivirals⁴ as well as antitumor agents. There are several reports of multi-component reactions for synthesis of *dihydropyrimidines*, for example the works reported by Hantzsch⁵ which described preparation 1,4-dihydropyridine three of using components (acetoacetic ester, benzaldheide and ammonia or ammoniumsalts) coupling reaction in refluxin ethanol. Other reports made by Bignelli⁶ showed the synthesis of dihydropyrimidines derivatives using ethyl acetoacetate, benzaldheyde and urea. Additionally, recently the dihydropyrimidin-2(1H)-one was synthesized using the three component system (urea/thiourea. ethvl acetoacetate/acetyl acetone) in presence of phosphorus pentoxide⁷. Another works showed the synthesis of 3,4dihydropyrimidin-2-(1H)-ones under solvent-free conditions using ruthenium(III) chlorid-catalyzed⁸. In addition, Kappe9 and coworkers showed a highly versatility solid-phase synthesis of biofunctional 4-aryl-3.4-dihydropyrimidines using resin-bound isothiourea building blocks and multidirectional resin cleavage. In

addition, Shirini and coworkers¹⁰ display that Fe(HSO₄)₃ as an efficient catalyst for the preparation of 3.4*dihydropyrimidin-2(1H)-ones* using the three component system (β -keto ester, benzaldheyde and thiourea). Another studies made by Salehia¹¹ and coworkers showed the synthesis of dihydropyrimidinones using aldheyde-derivatives, *dicarbonyl* compounds and urea or thiurea in presence of diammonium hydrogen phosphate. All these experimental data show several protocols for synthesis of dihydropyrimidine-derivatives, nevertheless, the use of expansive reagents requires of special conditions. In this work our initial design included an easy synthesis of a steroid-dihydropyrimidine derivative. The route involve preparation of progesteronedihvdropyrimidine derivative (4) using the three components system (4-pregnene-3,20-dione, benzaldheide and thiourea in presence of chloridric acid such as catalyst.

Experimental

General methods

Progesterone (4-*pregnene-3,20-dione*) and the other compounds evaluated in this study were purchased from Sigma-Aldrich Co., Ltd. The melting points for the different compounds were determined on an Electrothermal (900 model). Ultraviolet spectroscopy (UV) was carried out in dry methanol on a Perkin-Elmer model 552 spectrophotometer and infrared spectra (IR) was recorded using KBr pellets on a Perkin Elmer Lambda 40 spectrometer. ¹H and ¹³C NMR spectra were recorded on a Varian VXR-300/5 FT NMR spectrometer at 300 and 75.4 MHz in CDCl3 using TMS as internal standard. EIMS spectra were obtained with a Finnigan Trace GCPolaris Q. spectrometer. Elementary analysis data were acquired from a Perkin Elmer Ser. II CHNS/0 2400 elemental analyzer.

Synthesis of 11a,13a-dimethyl-phenyl-1-[1-(6-phenyl-2-thioxo-1,2,3,6-tetrahydro-pyrimidin-4-yl)-ethyl]-1,2,3,3a,3b,4,5,5a,7,9,10,10a,11,11a,11b,12,13,13aoctadeca-hydro-1*H*-7,9-diaza-indeno[5,4*a*]anthracene-8-thione.

A solution of *progesterone* 100 mg (0.31 mmol), *thiourea* 71 mg (0.93 mmol), *benzaldehyde* 60 μ L (0.93 mmol) in *ethanol* 10 mL was stirring by 10 minutes at room temperature. After *chloridric acid* 1 mL was added and the mixture was stirring by 48 hours at room temperature. The reaction mixture was evaporated to a smaller volume, diluted with water, and extracted with chloroform. The organic phase was evaporated to dryness under reduced pressure, the residue was purified by crystallization from methanol:water (3:1) yielding 70 % of product, mp 150 °C; UV (MeOH) λ_{max} (log ε) 219

(3.33), 278 (2.56) nm; IR V_{max} 3430, 1610 cm-1; $^1\mathrm{H}$ RMN CDCl3 (300 MHz) δ_H; 0.61 (3H, s, 22-CH₃), 0.72-0.80 (2H, m), 0.86 (3H, s, 32-CH₃), 1.05 (3H, s, 23-CH₃), 1.08-1.16 (2H, m), 1.21-1.33 (2H, m), 1.44-1.66 (5H, m), 1.70-1.81 (2H, m), 1.97 (1H, m), 2.09 (1H, m), 2.20-230 (2H, m), 2.97-3.03 (2H, m), 4.48 (1H, m), 4.75 (1H, m), 5.14 (1H, d, -CH=C, dihydropyridimidine-ring), 6.27 (1H, d, -CH=C, steroid-ring nucleus), 7.23-7.33 (6H, m, phenyl-rings), 7.50-7.55 (4H, m, phenyl-rings), 7.70 (4H, m, -NH). ¹³C RMN 13 CDCl₃ (75.4 MHz) δ_C; 12.26 (C-23), 14.70 (C-32), 15.56 (C-22), 24.25 (C-15), 25.35 (C-21), 28.28 (C-20), 28.53 (C-14), 34.33 (C-13), 37.37 (C-8), 37.90 (C-7), 37.90 (C-12), 39.94 (C-24), 40.05 (C-13), 40.15 (C-16), 48.47 (C-17), 53.96 (C-11), 55.53 (C-19), 57.25 (C-18), 63.24 (C-6), 63.69 (C-35), 108.22 (C-34, -C=C-, dihydropyridimidine-ring), 110.02 (C-10, CH=C, steroid-ring nucleus), 114.43 (C-1, C=C), 127.00 (C-42), 127.79 (C-40,C-44), 128.20 (C-28), 128.59 (C-41, C-43), 128.90 (C-27, C-29), 129.69 (C-26, C-30), 141.43 (C-25), 142.18 (C-2, C=C, dihydropyridimi- dine-ring), 145.84 (C-39), 148.26 (C-9, steroid-ring nucleus), 151.25 (C-33, -HN-C=C), 165.02 dihydropyridimidine-ring), (C-4, 177.02 (C-37, dihydropyridimidine-ring). EI-MS, m/s 634.93 (M⁺, 17), Anal. Calcd for C₃₉, H₄₆, N₄, S₂; C, 73.77, H, 7.30, N, 8.82, S, 10.10 Found: C, 73.51, H, 7.61, N, 8.78.

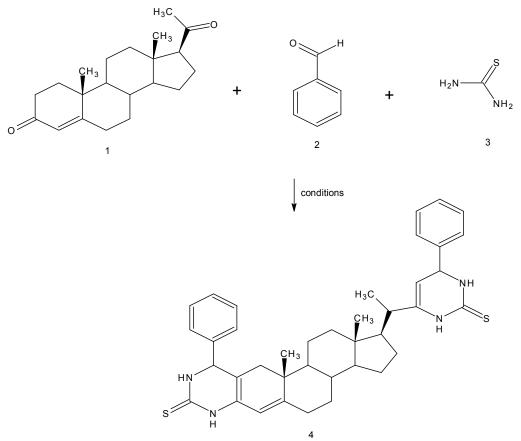


Figure 1. Synthesis of *progesterone-dihydropyrimidine derivative* (4) using the three components system (4-pregnene-3,20-dione (1), *benzaldheide* (2) and *thiourea* (3). Conditions: HCl/Ethanol

Results and Discussion

In this study the *progesterone-dihydropyrimidine* derivative was synthesized using the three component system (*4-pregnene-3,20-dione* [1], *benzaldheide* [2] and *thiourea* [3]) in presence of *chloridric acid* as catalyst. It is important to mention that many procedures for the formation of *dihydropyrimidine derivatives* are known in the literature. The most widely practiced method employs boric acid¹², silica sulfuric acid¹³, poly(4-vinylpyridine-codivynylbenzene)-Cu(II) complex¹⁴, H₂SO₄¹⁵, silica triflate¹⁶ and phosphorus pentoxide¹⁷. Nevertheless, despite its wide scope, the former protocols suffer from several drawbacks; some reagents have a limited stability and its preparation can be dangerous.

The results indicate that the ¹H NMR spectrum of *progesterone-dihydropyrimidine derivative* showed a signal at 0.61, 0.72 and 1.05 ppm for methyls present in the *steroid-rings nucleus*; at 5.14 and 6.27 ppm for methylens involved in *pyrimidine-ring and steroid-rings nucleus*. Additionally, there are several chemical shifts (723-733 and 750-7.55 ppm) corresponding to protons in

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aromatic-rings. Finally, spectra display a chemical shift similar at 7.70 ppm for -NH group (pyrimidine-ring). On the other hand, ¹³C NMR spectra displays chemical shifts at 12.26, 14.70 and 15.56 ppm for the carbons of the methyls groups presents in the steroid-rings nucleus. The chemical shift of the methylene joined to *pyrimidinering* at 108.22 ppm (-C=C-, *dihydropyridimidine-ring*) and 110.02 ppm (-C=C-, steroid-ring nucleus) were found. At down field there are several signals (127-141.43 and 145.84 ppm) corresponding to the carbons of aromatic-rings. Finally, several characteristic signals at 142.18 (-C=C-, dihydropyridimidine-ring), 148.26 for steroid-ring nucleus and 151.25 for carbon bound to nitrogen involved on pyrimidine-ring and at 165.02 and 177.02 for carbon bound to sulphur atom involved on *pyrimidine-ring*. Additionally, the mass spectra display a molecular ion of m/z 634.93 (M⁺, 17) which confirm the structure of 4. In conclusion in this work, we reported an easy procedure for the synthesis of progesteronedihydropyrimidine derivative.

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