



International Journal of ChemTech Research CODEN( USA): IJCRGG ISSN : 0974-4290 Vol.1, No.2, pp 195-198, April-June 2009

# Development and Validation of RP-HPLC for the Pantoprazole Sodium Sesquihydrate in Pharmaceutical dosage forms and Human Plasma

# Prasanna Reddy.Battu and N.Kiran Kumar Reddy

Department of Quality Control and Quality, Assurance, Smilax labs ltd, Jeedimetla,

Hyderabad-500055, A.P, India. Email: drbpkreddy@gmail.com

**Abstract:** A simple, selective, accurate high Performance Liquid Chromatographic (HPLC) method was developed and validated for the analysis of Pantoprazole sodium. Chromatographic separation achieved isocratically on a  $C_{18}$  column [Use Inertsil  $C_{18}$ , 5µ, 150 mm x 4.6 mm] utilizing a mobile phase of acetonitrile/phosphate buffer (70:30, v/v, pH 7.0) at a flow rate of 0.8 ml/min with UV detection at 260 nm. The retention time of Pantoprazole sodium was 2.017 min. The method is accurate (99.15-101.85%), precise (intra-day variation 0.13-1.56% and inter-day variation 0.30-1.60%) and linear within range 0.1-30µg/ml (R<sup>2</sup>=0.999) concentration and was successfully used in monitoring left over drug. The detection limit of Pantoprazole at a signal-to-noise ratio of 3 was 1.80ng/ml in human plasma while quantification limit in human serum was 5.60 ng/ml. The proposed method is applicable to routine analysis of Pantoprazole in pharmaceutical formulations as well as in human plasma samples.

Keywords: Pantoprazole, RP-HPLC, Validation, Human blood samples, Dosage forms

## Introduction

The proton-pump inhibitor pantoprazole inhibit gastric acid by blocking the  $H^+/K^+$ -adenosine triphosphatase enzyme system (the proton pump) of the gastric parietal cell<sup>1</sup>. It is used for short-term treatment of erosion and ulceration of the esophagus<sup>2</sup>. The pantoprazole oral dosage forms are supplied in enteric-coated tablets.Different analytical methods are reported in the literature for the assay of pantoprazole in dosage forms and in biological fluids including spectrophotometry<sup>3-8</sup>, TLC<sup>9</sup>, HPTLC<sup>10-12</sup>. Pantoprazole sodium is chemically Sodium 5-(difluoromethoxy)-2-[[(3,4-dimethoxy-2pyridinyl) methvl] sulfinyl]-1*H*-benzimidazole sesquihydrate. It has an empirical formula of C<sub>16</sub>H<sub>15</sub>F<sub>2</sub>N<sub>3</sub>O<sub>4</sub>S and molecular weight of 383.37. The aim of this work was to develop new and validated, simple and reproducible RP-HPLC method allowing the estimation of in dosage forms and human plasma samples.

## Experimental

A High Performance Liquid Chromatograph system, with LC solutions data handling system (Shimadzu-LC2010) with an auto sampler was used for the analysis. The data

was recorded using LC 2010 solutions software. The purity determination performed on a stainless steel column 150 mm long, 4.6 mm internal diameter filled with Octadecyl silane chemically bonded to porous silica particles of 5 $\mu$ m diameter (Inertsil C<sub>18</sub>, 5 $\mu$ , 150 mm x 4.6 mm, make: Shimadzu ltd, Japan) with the mobile phase containing acetonitrile and phosphate buffer in the ratio of 70:30 (v/v pH 7.0) at ambient temperature. Flow rate was kept at 0.8 ml/min and the elution was monitored at 260 nm.

## **Materials and Chemicals**

Pantoprazole sodium working standard, used from Smilax Laboratories Limited. For the estimation of Pantprazole sodium in bulk and commercial formulations of pantoprazole sodium brand (Pantin-20, Genex laboratories), 20 tablets were obtained from retail pharmacies. Each tablet was labeled contain 20 mg of pantoprazole sodium and had an expiry of not less than 365 days at the time of study. HPLC grade Sodium dihydrogen phosphate (NaH<sub>2</sub>PO<sub>4</sub>) .Disodium hydrogen phosphate (Na<sub>2</sub>HPO<sub>4</sub>), Acetonitrile- procured from Merck, India. High pure water was prepared by using Millipore Milli Q plus purification system.

#### **Preparation of mobile phase:**

Mobile phase was prepared by mixing 700 ml of acetonitrile with 300 ml of phosphate buffer and its pH adjusted to 7.0. The mobile phase was sonicated for 15 min and then it was filtered through a 0.45  $\mu$  membrane filter paper.

#### Preparation of stock and standard solutions:

Accurately weighed 25 mg of test sample into a clean dry 50 ml volumetric flask, dissolve and dilute to the mark with mobile phase. Mark this solution as sample solution. This solution contains 0.5 mg/ml of sample. Qualified working standard of Pnatoprazole sodium is used to carry out validation exercise. The potency of working standard is 99.75 %. With the optimized chromatographic conditions, a steady baseline was recorded, the standard solution was injected and the chromatogram was recorded. This procedure was repeated for the sample solution.

## Method validation

The method was validated for the parameters like specificity, range and linearity, limit of detection (LOD), limit of quantitation (LOQ), accuracy, and precision. In addition, system suitability parameters were also calculated. To demonstrate specificity in the presence of excipients used in formulation, pantoprazole sodium was spiked (at approximately 25 µg/ml) in drug product, chromatogram was observed and compared with that of raw material. To evaluate the linearity, the LOD and LOQ of the method in reference drug and in serum, different serial dilutions (0.0980, 0.190, 0.80, 1.50, 3.10, 6.30, 12.50 and 25 µg/ml) were prepared from the standard stock solutions in 25 ml volumetric flasks and volume made up with diluent which is mixture of 70:30 acetonitrile & methanol. The samples were injected (10 µl) and signals from the samples were recorded at 2.02 minute which were compared with those of blank. LOD and LOO values were calculated as signal-to-noise ratio of 3:1 and 10:1 respectively. To determine accuracy of the method, working standard of pantoprazole sodium was prepared in triplicate at three concentration levels (10, 20 and 25 µg/ml) and analyzed. Repeatability of the method was checked by analyzing six replicate samples of pantoprazole sodium (at the 100% concentration level) and calculating relative standard deviation (%RSD). To determine intermediate precision, standard solutions of pantoprazole sodium at eight concentration levels were analyzed three times within the same day (intra-day variation) and three other days (inter-day variation).

## Assay in formulations

In case of marketed formulations, five accurately weighed tablets were crushed to a fine powder and an amount equivalent to 10 mg of pantoprazole sodium was added into different 100 ml volumetric flasks and volume was made up with acetonitrile and methanol mixture. The samples were filtered through a 0.45-µm-membrane filter; different serial dilutions (3.10, 6.20, 12.40,

196

25μg/ml) were made from this solution in 25 ml volumetric flask and were injected for HPLC analysis. **Assay in serum** 

One volume of plasma was de-proteinated by nine volume of acetonitrile and filtered through 0.45  $\mu$ m Millipore filter paper that was used to make serial dilution of Rabeprazole (0.0970  $\mu$ g/ml to 25  $\mu$ g/ml). Three replicates of each dilution were injected to HPLC system and linearity was evaluated. Repeatability of the method was checked by analyzing six replicate samples of pantoprazole sodium (at the 100% concentration level) and calculating relative standard deviation (%RSD).

Table 1: Accuracy/recovery of Pantoprazolesodium

Parameters	Conc (µg/ml)	% Recovery	% RSD
Assay	10	96.02	1.70
(Spiking method)	20	101.54	4.80
	25	95.15	4.60
Assay	6.30	99.18	0.5
	12.40	100	1.5
	25	99.99	0.3
Assay (in serum)	12.40	100	0.5
	6.20	100	1.2
	3.10	100	0.7

Concen tration (µg/ml)	Assay in formulation		Serum
	Intra-day variation (%RSD)	Inter-day variation (%RSD)	Intra-day variation (%RSD
0.0980	0.13	0.90	4.12
0.190	0.35	0.33	0.08
0.80	0.40	1.60	2.70
1.50	0.85	1.08	3.20
3.10	0.26	0.20	1.15
12.5	1.65	0.70	0.40
25	0.25	1.08	3.30

#### **Results and discussion**

For validation of analytical methods, the guidelines of the International Conference on the Harmonization of Technical Requirements for the Registration of Pharmaceuticals for Human Use [ICH 1996] and [USP 2002] have recommended the accomplishment of accuracy tests, precision, specificity, linearity of the method

### System suitability

#### Prasanna Reddy.Battu.et al /Int.J. ChemTech Res.2009,1(2)

The HPLC system was equilibrated with the initial mobile phase composition, followed by 10 injections of the same standard. These 10 consecutive injections were used to evaluate the system suitability on each day of method validation.

The system suitability parameters including capacity factor >2, resolution>3 and asymmetric factor<2. All parameters were satisfactory with good specificity for the stability assessment of pantoprazole sodium. Theoretical plates of the column were >3000.

### Accuracy

The accuracy of an analytical method is the closeness of test results obtained by that method to true value. In case of the assay of a drug in a formulated product, accuracy may be determined by application of the analytical method to synthetic mixtures of the drug product components to which known amount of analyte has been added within the range of method. If it is not possible to obtain samples of all drug product components, it may be acceptable to add known quantities of the analyte to the drug product (i.e.,"to spike") (USP 2004). In our studies, the later technique was adopted and pantoprazole was spiked in drug product. The result of accuracy given in (Table-1) revealed that the method was found accurate for all above purposes.

#### Precision

Precision is the degree of reproducibility or repeatability of the analytical method under normal operating conditions (USP 2004). The method passed the test for repeatability as determined by %RSD of the area of the peaks of six replicate injections at 100% test concentration. The results of intra-and inter-day variation are shown in (Table 2).

#### **Range and Linearity**

The linearity of an analytical method is its ability to elict test results that are directly, or by a well-defined mathematical transformation, proportional to the concentration of analyte in samples within a given range (USP 2004). The linearity of the method was observed with in the expected concentration range demonstrating its suitability for analysis. The correlation coefficient  $(r^2)$ was found to be 0.999 and value of intercept was less than 25 of the response of 100% of the test concentration in all the cases indicating functional linear relationship between the concentration of analyte and area under the peak.

## Limits of Detection and Quantitation

The detection limit (LOD) is the lowest amount of an analyte in a sample that can be detected, but not necessarily quantitated, under the stated experimental conditions. It may be expressed as a concentration that gives a signal-to-noise ratio of 2:1 or 3:1 (USP 2004, ICH Q2B guidelines, 1996 1997, FDA, Guidance for Industry 2000)<sup>13, 14</sup>. The lower limit of detection for rabeprazole is 2.40ng/ml in reference material and formulation and 1.70ng/ml serum. Limit of Quantitation (LOQ) is the lowest amount analyte in a sample that can be determined with acceptable precision and accuracy under the stated experimental conditions. A signal-to-noise ratio of 10:1 can be taken as LOQ of the method (USP 2004). The LOQ values were found to be 8.15ng/ml for raw material, formulations and 5.70ng/ml for serum.

## Specificity

Specificity is the ability to assess unequivocally the analyte in the presence of components that may be expected to be present in the sample matrix (USP 2004). For demonstrating the specificity of the method for drug formulation the drug was spiked and the representative chromatogram (Figure-1). The excipiants used in different formulation products did not interfere with the drug peak and thus, the method is specific for pantoprazole. To further confirm the specificity of the method, UV scans of spiked drug were taken in the range 200-400nm and no significant change was found by comparing the absorbance of pure drug and spiked drug at the analytical wavelength of drug.



Figure-1 Chromatogram of Pntoprazole sodium

## **References:**

- Ritter j. m., Lewis L.D., Mant T.G.K., "A Textbook of Clinical Pharmacology", 4<sup>th</sup> ed., Arnold LTD London, 1999, p. 365.
- Ewin K.J., "Goodman & Gilman's. The Pharmacological Basis of Therapeutics", 10<sup>th</sup> ed., McGraw-Hill Inc., London, 2001, p. 1007.
- Ozaltin N., Kocer A., J. Pharm. Biomed. Anal., 1997, 16, 337-342
- 4. Sastry C.S.P., Naidu P.Y., Murty S.S.N., Talanta, 1997, 44, 1211-1217.
- 5. Meyyanathan S.N., Raj J. R. A., Suresh B., Indian Drugs, 1997, 34, 403-406.
- 6. Moustafa A.A. M., J.Pharm. Biomed Anal., 2000, 22, 45-58.
- Wahbi A. A. M., Abdel-Razak O., Mahgoub Gazy A. A. H., Moneeb M.S., J.Pharm. Biomed Anal., 2002, 30, 1133-1142.

- Salama F., Abasawy N. E.I., Abdel Razeq S.A., Ismail M.F., Fouad M.M., J.Pharm. Biomed Anal., 2003, 32, 1019-1027.
- EI Sherif Z.A., Mohamed A. O., EI-Bardeicy M.G., EI-Tarras M. F., Spectroscopy Lett., 2005, 38, 77-93.
- 10. Renger B., J. AOAC. Int., 1993, 76, 7-13.
- 11. Argekar A.P., Kunjir S.S., J Planar-Chromator. Mod., 1996, 9, 296-299.
- Pandya K.K., Mody V.D., Satia M.C., Modi I. A., Modi R.I., Chakravarthy B. K., Gandhi T.P., J Chromatog. B. Biomed. App 1997, 693, 199-204.
- 13. ICH Q2B: Validation of Analytical Procedures: Methodology, May (1997).8.
- 14. International Conference on the Harmonization of Technical Requirements for the Registration of Pharmaceutical for Human Use (ICH) Q2B (1996). Validation of Analytical Procedures, Methodology.

\*\*\*\*\*